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In the space below write the unabbreviated electron configurations of the following elements: 1. sodium, 2. iron, 3. bromine, 4. barium, 5. neptunium, 6. cobalt, 7. silver, 8. tellurium. In the space below write the abbreviated electron configurations of the following elements: 1. sodium, 2. iron, 3. bromine, 4. barium, 5. neptunium, 6. cobalt, 7. silver, 8. tellurium.

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Quiz Form A Please write out the electron configuration of the following elements. Show both the order of filling and by energy level. 1. W #74, 2. Gd #64. How many valence electrons does each of the above have? 3. Please write out the following electron configurations using the short cut method. Bi #83, Y +3 #39. 4. Please circle the atom with the larger atomic radius (if they are equal just write =) K #19 ...

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ELECTRON CONFIGURATION (LEVEL ONE) Electrons are distributed in the electron cloud into principal energy levels (1, 2, 3, ...), sublevels (s, p, d, f), orbitals (s has 1, p has 3, d has 5, f has 7) and spin (two electrons, allowed per orbital). Example: Draw the orbital diagram of sodium (atomic #11). Answer:  $1s^2 2s^2 2p^6 3s^1$ . Draw the orbital diagram of...

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