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Honour Chemistry: Unit 4 Practice Test: Thermochemistry and Nuclear ChemistryPart A: Multiple Choice and Numerical Response (1 mark each for Multiple Choice) (2 marks each for Numerical Response)  $E_{\text{bind}} = \sum mc^2$   $c = \text{speed of light}$  (3.00 10<sup>8</sup> m s<sup>-1</sup>) 1 eV = 1.69 10<sup>-19</sup> J (1 kg = 6.022 10<sup>26</sup> amu , or 1 g = 6.022 10<sup>23</sup> amu = 1 mole amu) 1 MeV = 1.69 10<sup>-16</sup> J (1 Neutron = 1.008665 amu) (1 Proton = 1.007825 amu)

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Honour Chemistry Practice Test: Unit 2 (Part 1): Matter as Solutions and Gases 1 atm = 760 mm Hg = 760 torr = 101.325 kPa  $T_K = T_C + 273.15$  STP = 22.4 L mol @ 0.00C & 101.325 kPa (1 atm) SATP = 24.8 L mol @ 25.0C & 100. kPa L atm L kPa m m nsolute  $R = 0.0821$  or 8.314 D= n= C= K mol K mol V M Vsolution  $P_1V_1 = P_2V_2$   $r_1M_2 = n_1P_1 = C_1V_1 = C_2V_2$   $PV = nRT$  = = ?1 = = n1T1 n2T2 r2 M1 nTotal PTotal PTotal = P1 +...

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Practice Test 7, solutions belowProblem 11. (p & q) A2. (p v q) A3. Show: rProblem 21. (p ? q) ? (r ? s) A2. p v q A3 (q & p) A4. Show: (r v s)Problem 31. (p v q) v ( r v s) A2. r A3. (t ? s) A4. Show p ?q Problem 11. (p & q) A2. (p v q) A3. Show: r4. p & q &, 25. p & q , 46. p & q , 57. r !, 1,6Problem 21. (p ? q) ? (r ? s) A2. p v q A3 (q & p) A4. Show: (r v s)5. (r v s) A1P6. p ? q ?v, 27. q v ...

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