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Algebra 2 P Chapter Practice Test Name: Please show your work. You are allowed only a basic calculator. This is last years test.1. Perform the indicated operations and simplify 2a)  $(5x^2 - 7y)$  b)  $(2x + 7)(3x^2 - 8x + 1)$ 2. Factor each expression completely a)  $x^2 - 3x - 40$  b)  $2x^3 + x^2 - 6x - 3$  c)  $x^4 + 5x^3 + 6x^2$  d)  $2(a + b)^2 + 5(a + b) - 33$ . a) Write  $?x < 8$  in interval notation and...

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### [UNIT 4 PRACTICE TEST](#)

Honour Chemistry: Unit 4 Practice Test: Thermochemistry and Nuclear ChemistryPart A: Multiple Choice and Numerical Response (1 mark each for Multiple Choice) (2 marks each for Numerical Response)  $E_{\text{bind}} = ??mc^2$   $c = \text{speed of light}$  (3.00 10<sup>8</sup> m s)  $1 \text{ eV} = 1.69 \cdot 10^{19} \text{ J}$  (1 kg = 6.022 10<sup>26</sup> amu , or 1 g = 6.022 10<sup>23</sup> amu = 1 mole amu)  $1 \text{ MeV} = 1.69 \cdot 10^{16} \text{ J}$  (1 Neutron = 1.008665 amu) (1 Proton = 1.007825 amu)

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### [UNIT 2 \(PART 1\) PRACTICE TEST](#)

Honour Chemistry Practice Test: Unit 2 (Part 1): Matter as Solutions and Gases  $1 \text{ atm} = 760 \text{ mm Hg} = 760 \text{ torr} = 101.325 \text{ kPa}$   $TK = TC + 273.15$   $STP = 22.4 \text{ L mol @ } 0.00\text{C} \ \& \ 101.325 \text{ kPa}$  (1 atm)  $SATP = 24.8 \text{ L mol @ } 25.0\text{C} \ \& \ 100. \text{ kPa}$   $L \text{ atm} \ L \text{ kPa} \ m \ m \ n \text{ solute} \ R = 0.0821 \text{ or } 8.314$   $D = n = C = K \text{ mol} \ K \text{ mol} \ V \ M \ V_{\text{solution}} \ P1V1 \ P2V2 \ r1 \ M2 \ n1 \ P1 \ C1V1 = C2V2 \ PV = nRT = = ?1 = = n1T1 \ n2T2 \ r2 \ M1 \ n_{\text{Total}} \ P_{\text{Total}} \ P_{\text{Total}} = P1 + \dots$

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